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the meaning of osmosis is movement of a solvent such as water through a semipermeable membrane as of a living cell into a solution of higher solute concentration that tends to equalize the concentrations of solute on the two sides of the membrane how to transform your marks into grade achievements with osmosis s unique study tools effortless learning organizing assessing and collaborating are the keys to high performance students and educators who use osmosis tell us they see higher grades and lower stress osmosis is an efficient enjoyable and social way to learn sign up for an account today don t study it osmose it osmosis is the movement of solvent from a region of lower solute concentration to a region of higher solute concentration through a semi permeable membrane what are the three types of osmotic conditions that affect living cells the three types of osmotic conditions include hypertonic isotonic and hypotonic osmosis the spontaneous passage or diffusion of water or other solvents through a semipermeable membrane one that blocks the passage of dissolved substances i e solutes the process important in biology was first thoroughly studied in 1877 by a german plant physiologist wilhelm pfeffer oct 1 2020 osmosis is when a substance crosses a semipermeable membrane in order to balance the concentrations of another substance in biology this is usually when a solvent such as water flows into or out of a cell depending on the concentration of a solute such as salt osmosis happens spontaneously and without any energy on the part of the cell jul 3 2019 osmosis is the process where solvent molecules move through a semipermeable membrane from a dilute solution into a more concentrated solution which becomes more dilute in most cases the solvent is water however the solvent may be another liquid or even a gas osmosis can be made to do work login osmosis the best learning experience possible osmosis empowers students with our comprehensive video library combined with powerful study tools school email password keep me signed in i forgot my password or facebook google don t have an account sign up have a school account start here osmosis is the movement of a solvent across a semipermeable membrane toward a higher concentration of solute in biological systems the solvent is typically water but osmosis can occur in other liquids supercritical liquids and even gases jan 22 2022 by definition osmosis is the movement of any solvent through a selectively permeable membrane into an area of higher solute concentration the result of which will be an equalizing of solute concentration on either side of the membrane there are many times in life where you have heard someone preach about the importance of balance

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